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The seminar of three XMas fairy tales

Denis T.

★★ 21-12-2015 ★★



 \star The saga of Histamine

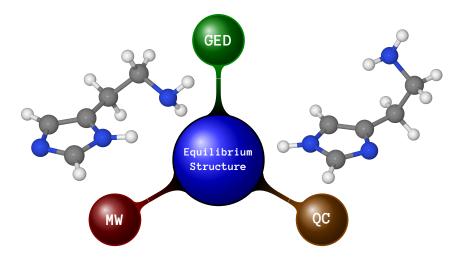
 \star The ballad of Pyrazinamide

 \star The legend of Qassandra



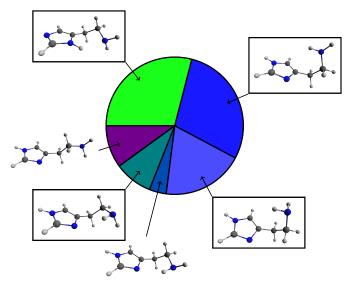
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The saga of Histamine



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Histamine: Vapor composition

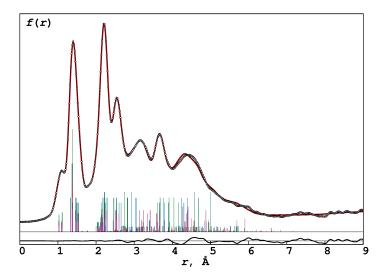


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Vogelsanger B. et al., *JACS*, 1991, **113**, 7864-7869 Godfrey P. et al., *JACS*, 1998, **120**, 10724-10732

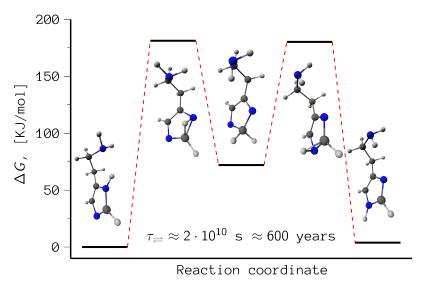
Histamine: GED+MW+QCReg refinement

 $R_f = 3.1\%$



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Histamine: intramolecular tautomerization

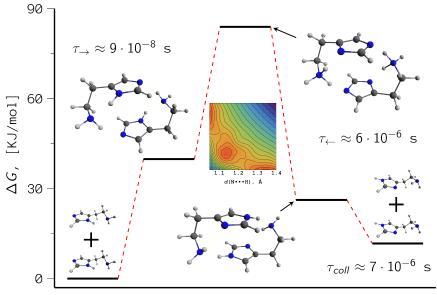


Borisov Y. et al., Bull. of the Acad. of Sc. of USSR, Chem., 1988, 37, 2504-2507

Nagy P. et al., J. Phys. Chem. B, 2005, 109, 22588-22602

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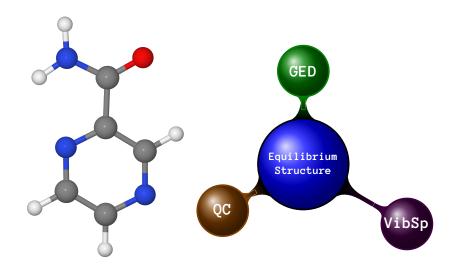
Histamine: intermolecular tautomerization



Reaction coordinate

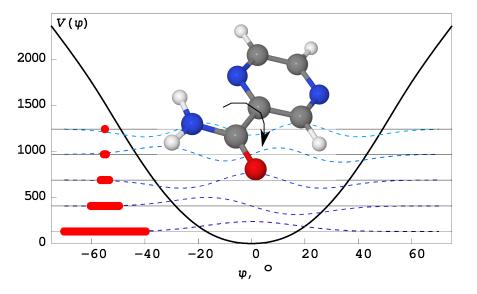
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The ballad of Pyrazinamide (PZA)



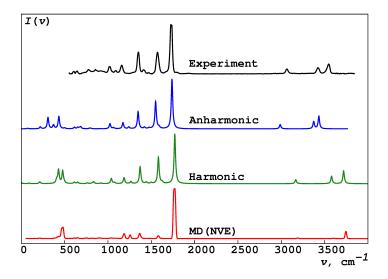
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PZA: large amplitude motion



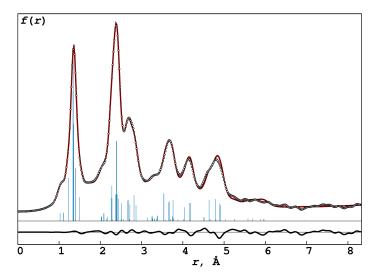
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PZA: comparison with VibSp



PZA: GED+QCReg refinement

 $R_f = 3.4\%$



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Qassandra

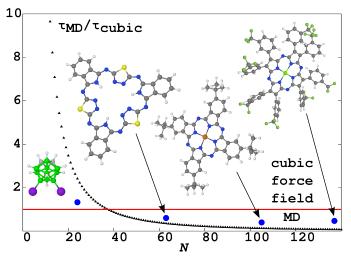
(QuAntum corrections to claSSical pArameters for gas electroN DiffRAction)



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Qassandra: Why MD?

MD vs. Cubic force field calculation



 Zhabanov Yu. et al., J. Mol. Struct., 2015, 1092, 114-112

 Pimenov O. et al., Struct. Chem., 2015, 26, 1531-1541

Qassandra: Problems of MD





"Flying Ice Cube" Effect

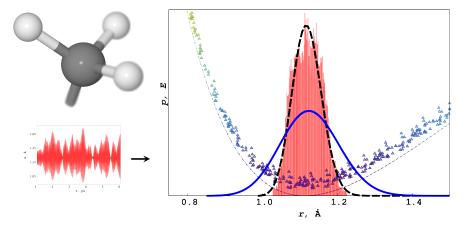
Harvey S. et al., J. Comput. Chem., 1998,

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19, 726-740

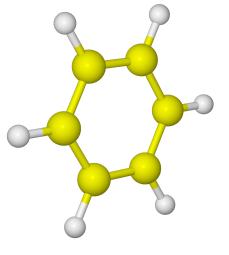
Classical Behavior

Qassandra: How does it work?



C-H term in ethane

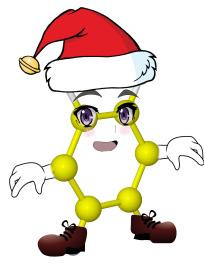
Qassandra: applicability check



Benzene

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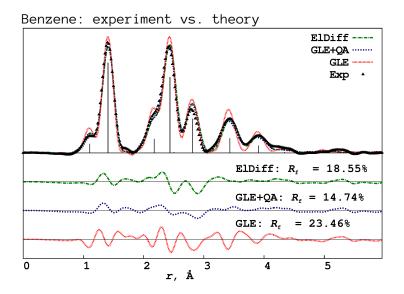
Qassandra: applicability check



XMas Benzene

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Qassandra: applicability check



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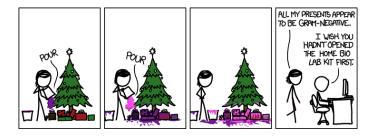
 \star MOAR r_e STRUCTURES! (Histamine and PZA)

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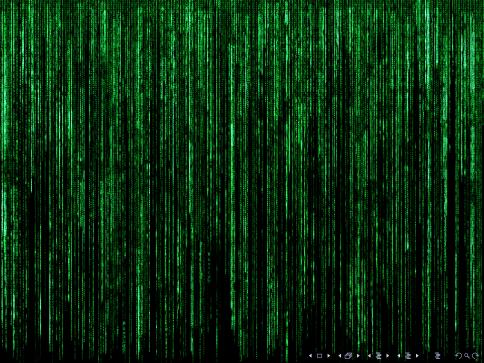
 \star The gas of histamine reaches tautomeric equilibrium

★ Qassandra is alive!

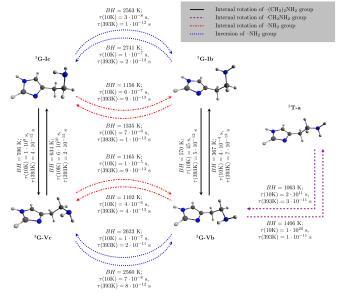
Thank You for the attention!



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Histamine: conformational equilibrium



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Types of systems in thermodynamics

Isolated system = NVE



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Heat exhange ··· X Matter exhange ··· X Types of systems in thermodynamics

Closed system = NVT



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Heat exhange ··· ✓ Matter exhange ··· ४ Types of systems in thermodynamics

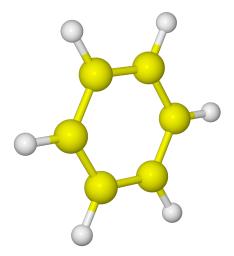
Open system = μVT



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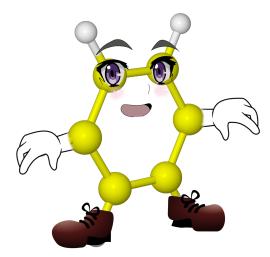
Heat exhange ··· ✓ Matter exhange ··· ✓

Molecular dynamics simulations (MD)



System = molecule

Molecular dynamics simulations (MD)



System = molecule

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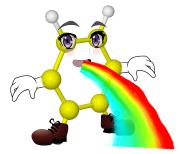
Classical motions on the PES

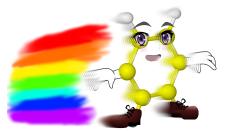


Classical motions on the PES with thermostat

Energy loss $(E_{kin}\downarrow)$

Energy gain $(E_{kin}\uparrow)$





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Problems in MD simulations: "Flying Ice Cube Effect"



Non-equilibrium distribution of energy for the degrees of freedom

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Solution: <u>better thermostats</u> Problem of the solution: limited availability

Problems in MD simulations: classical behavior

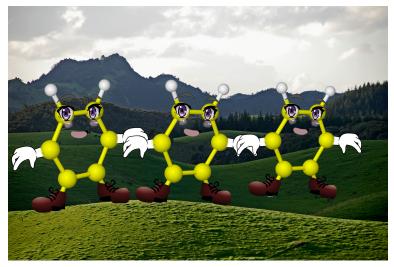


The lighter the atom - the more of quantum behavior

Solution #1: <u>PIMD</u> Problems of solution #1: computational costs increase, limited availability

Solution #2: <u>colored-noise thermostats</u> Problem of solution #2: limited availability

PIMD



Accounting quantum motions using path-integral approach