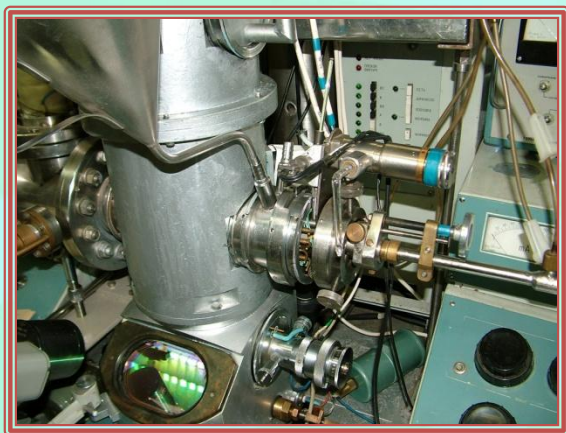




Combined gas-phase electron diffraction and mass spectrometry

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Ivanovo GED group

*(Ivanovo state university of chemistry and technology
+ Ivanovo state university)*





Foundation of Ivanovo GED group



Prof. George V. Girichev, the head of Ivanovo GED group



Prof. Sergey A. Shlykov

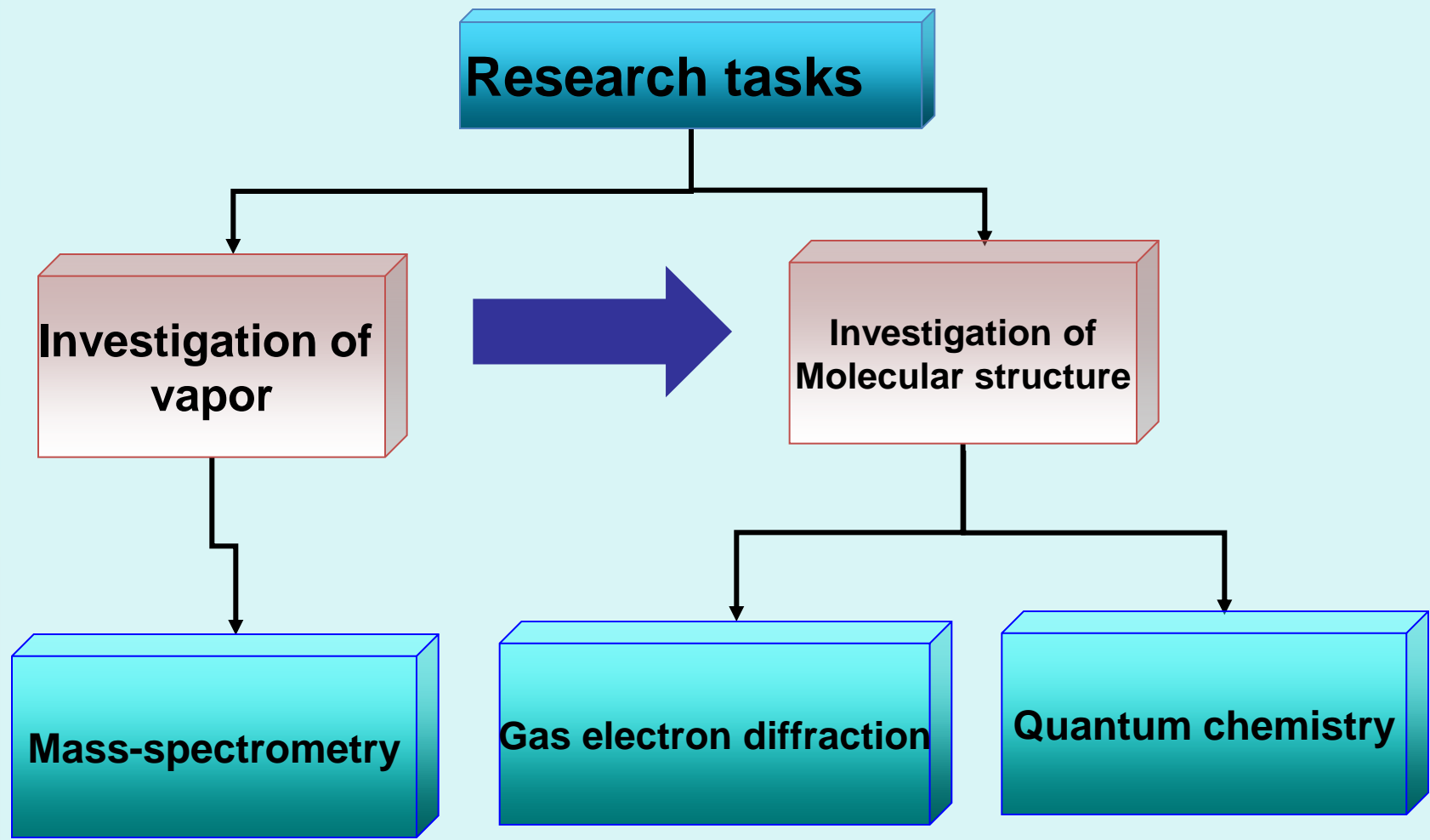


“electron diffractometer + mass spectrometer” (GED/MS)

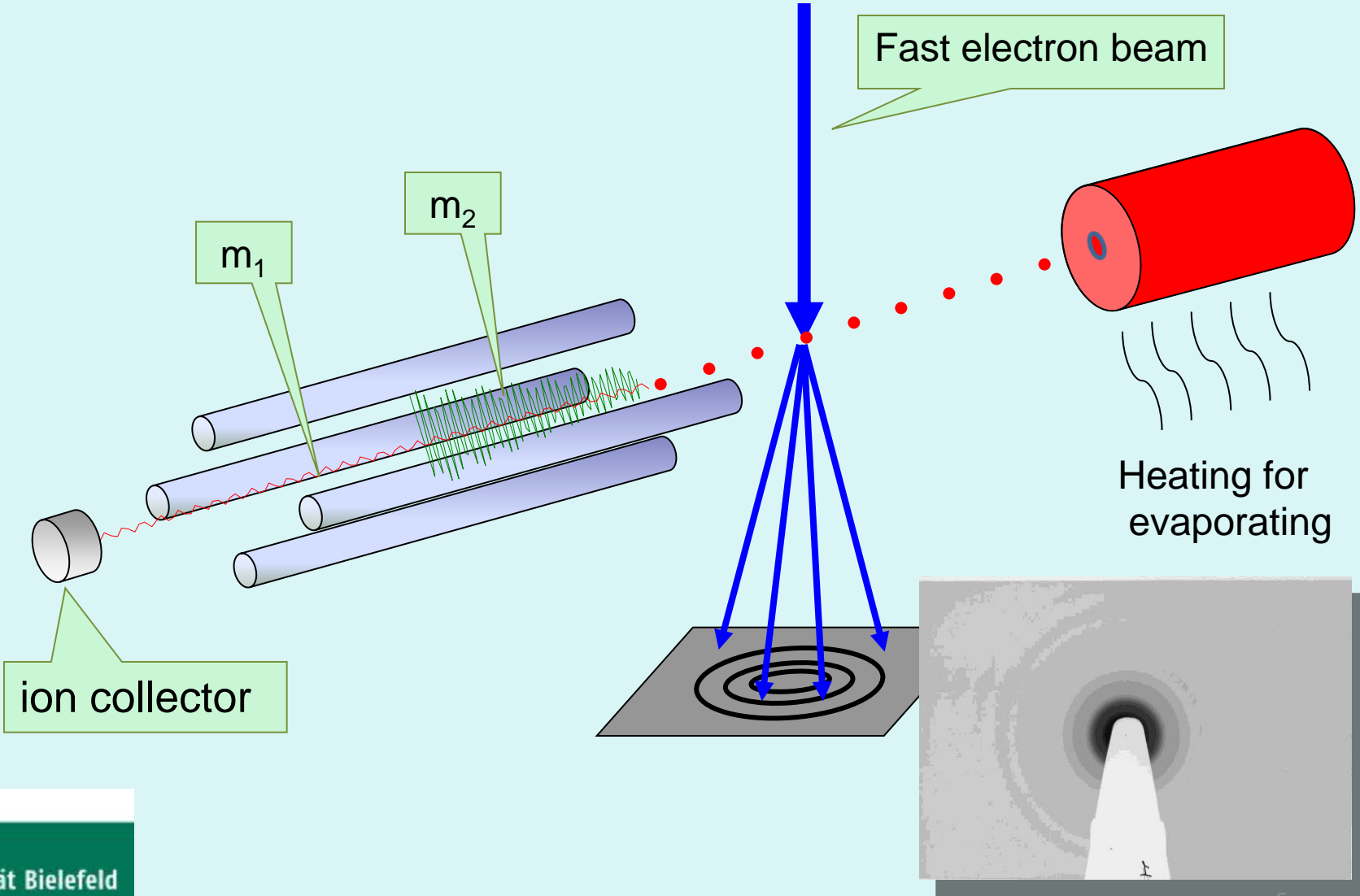
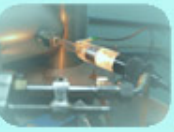
- 1. Girichev G. V., Utkin A. N., Revichev Yu. F., Prib. Tekh. Eksp. (Russian) 1984, N2, 187-190;*
- 2. Girichev G. V., Shlykov S. A., Revichev Yu. F., Ibid. (Russian) 1986, N4, 167-169*



Complex approach

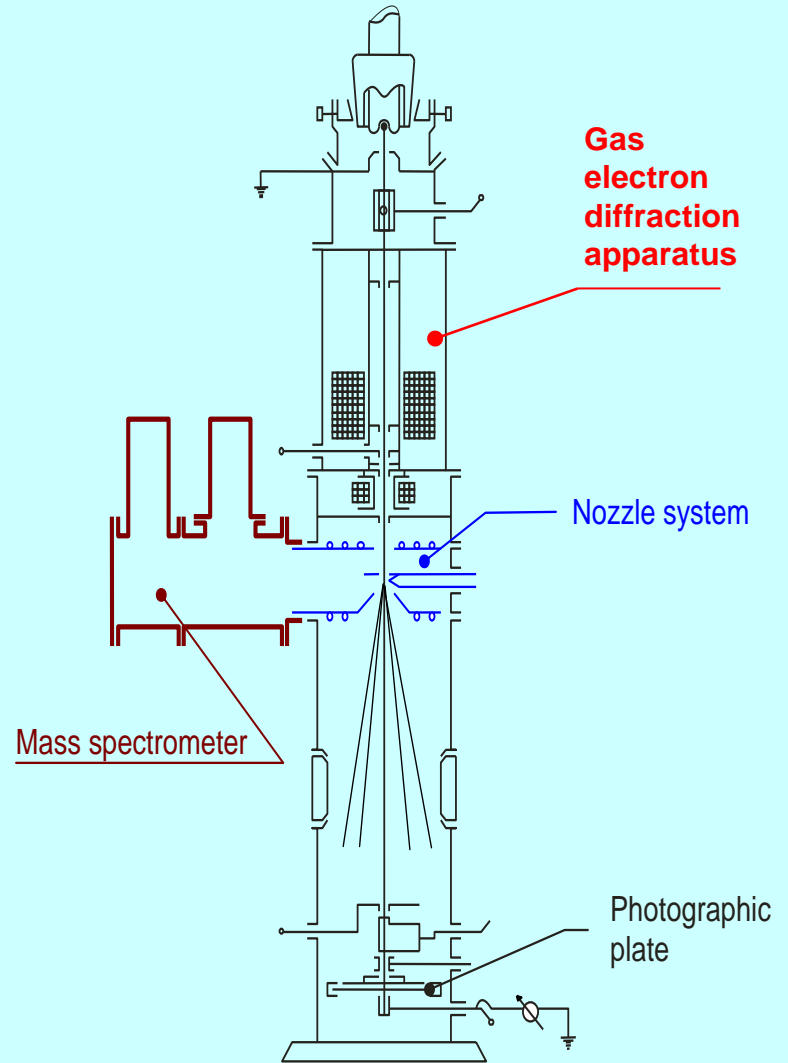
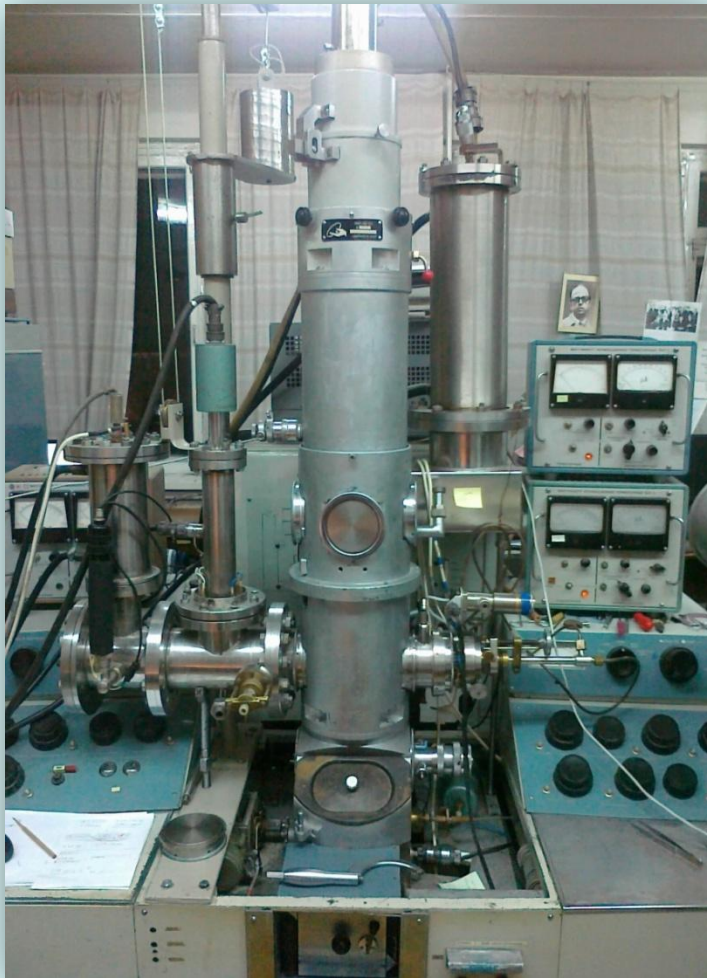


Synchronous gas electron diffraction and mass spectrometric experiment



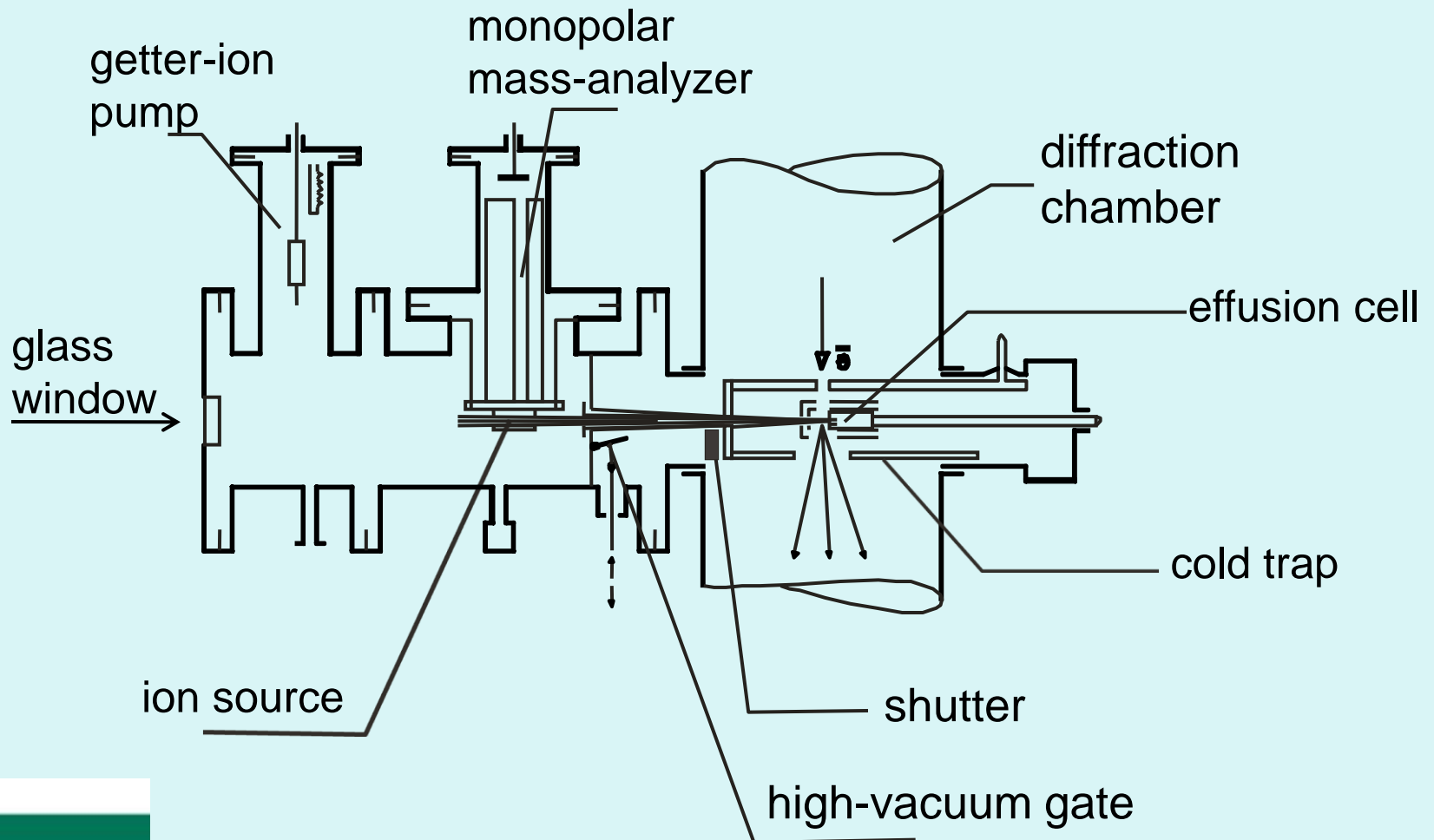


Ivanovo GED equipment





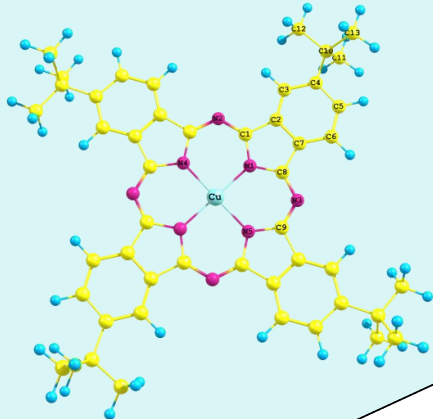
Ivanovo GED instrument/mass-spectrometer assembly



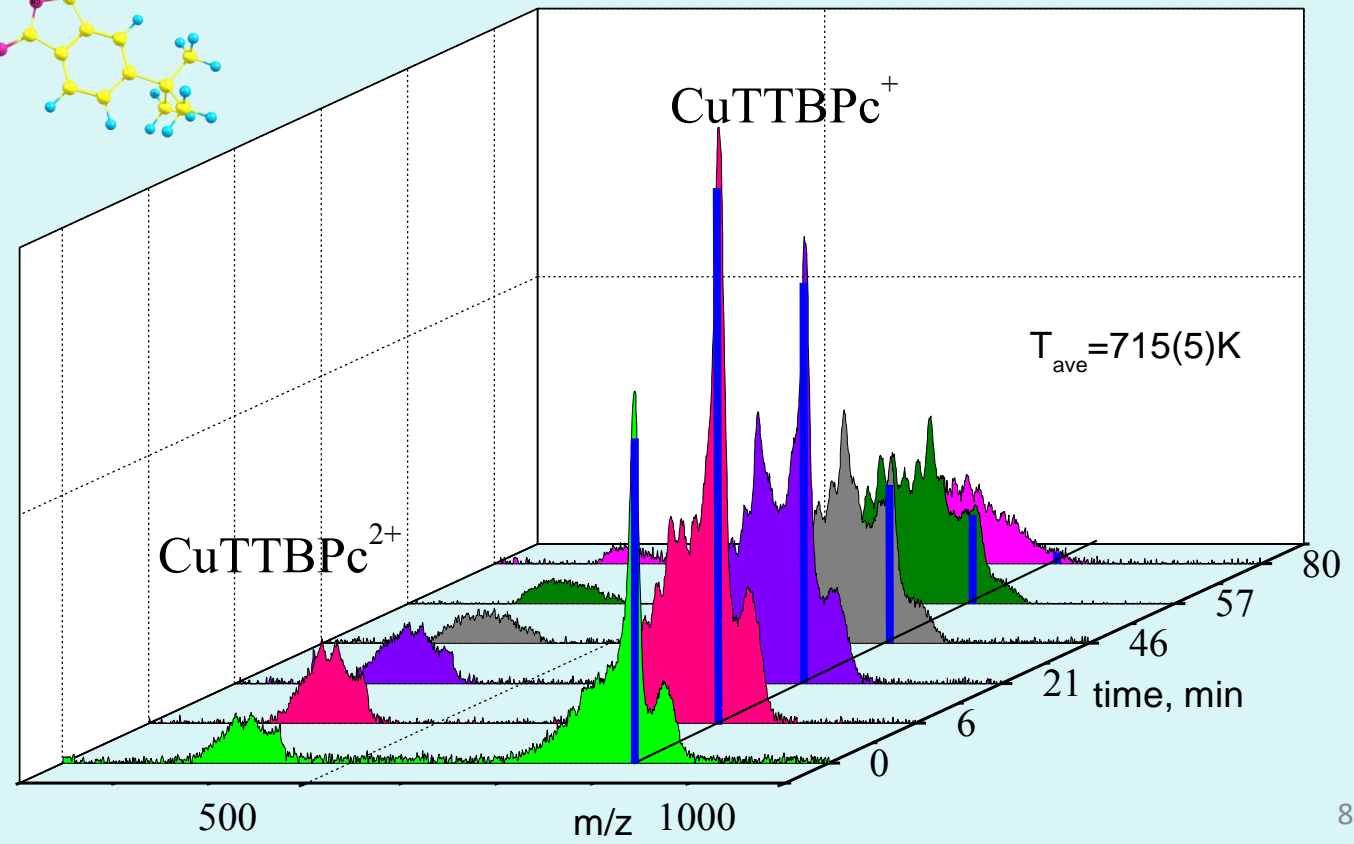


Mass-spectrometry is “eyes” of conventional electron diffraction

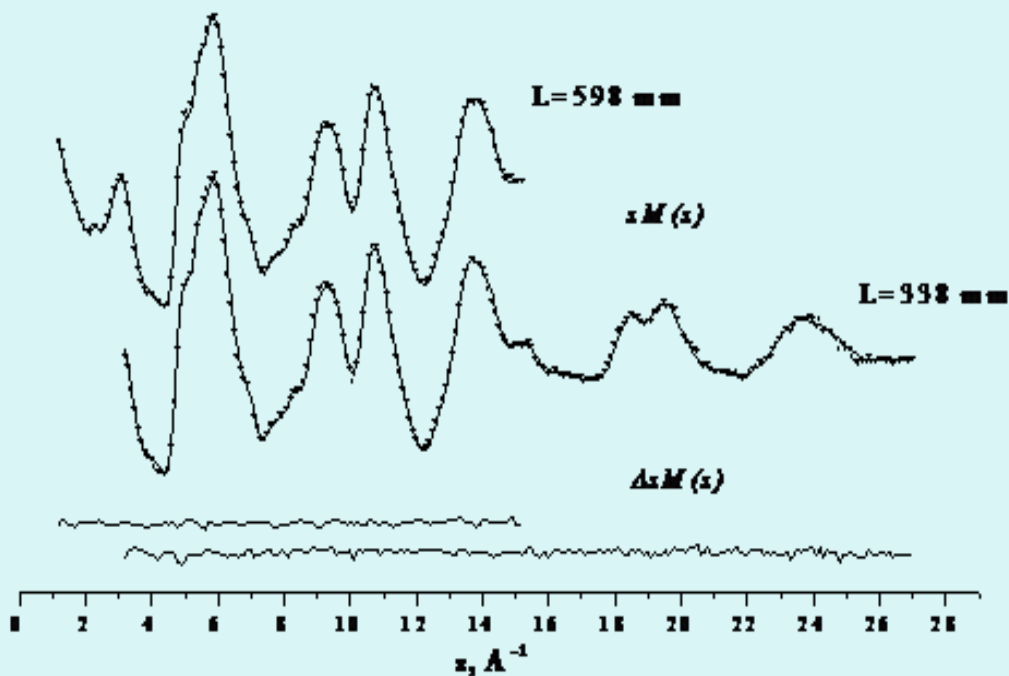
Copper (II) 2,9,16,23-tetra-*tert*-butyl-phthalocyanine (CuTTBPc)



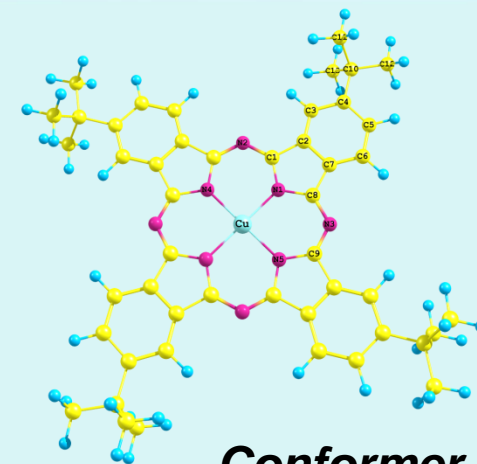
Time dependence of CuTTBPc vapor composition



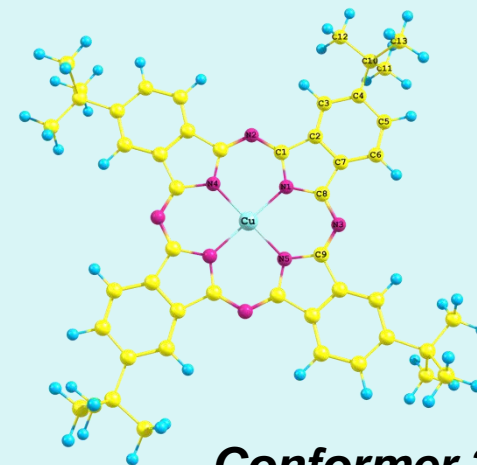
Molecular intensity functions $sM(s)$ of CuTTBPc (dots – experimental; solid line – theoretical)



$$R_{f2} / R_{f1} = 4.76 / 4.36 = 1.092 > 1.072^*$$



Conformer 1
(C_{4h} symmetry)



Conformer 2
(C_{4h} symmetry)

* Hamilton WC (1965) Acta Cryst 18:502

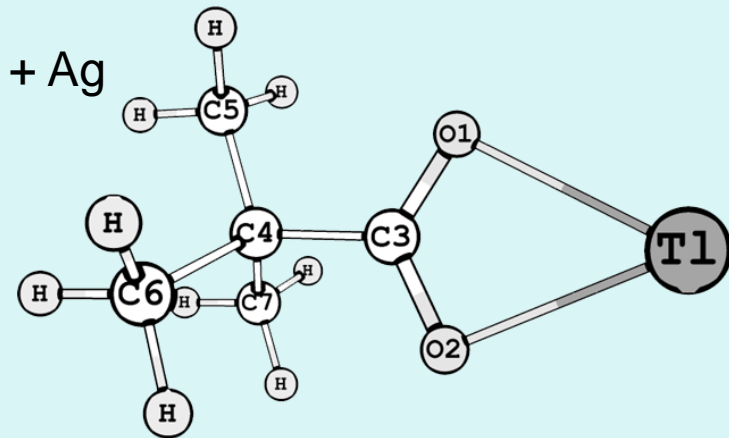
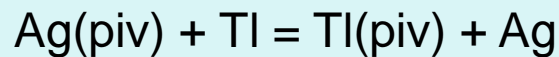
Even if all is not well, we smile...



Literature overview

N.N. Kamkin, L.G. Kuz'mina, D.B. Kayumova, N.G. Yaryshev, I.A. Dementiev, and A.S. Alikhanyan. Synthesis and Thermodynamic and Structural Characteristics of Thallium(I) Pivalate. Russian Journal of Inorganic Chemistry, 2012, Vol. 57, No. 9, pp. 1267–1271.

Thallium pivalate was synthesized and studied by mass spectrometry. It was shown that the set of characteristics makes thallium pivalate suitable as a precursor for the CVD preparation of oxide films and oxide materials.





GED/MS experiment

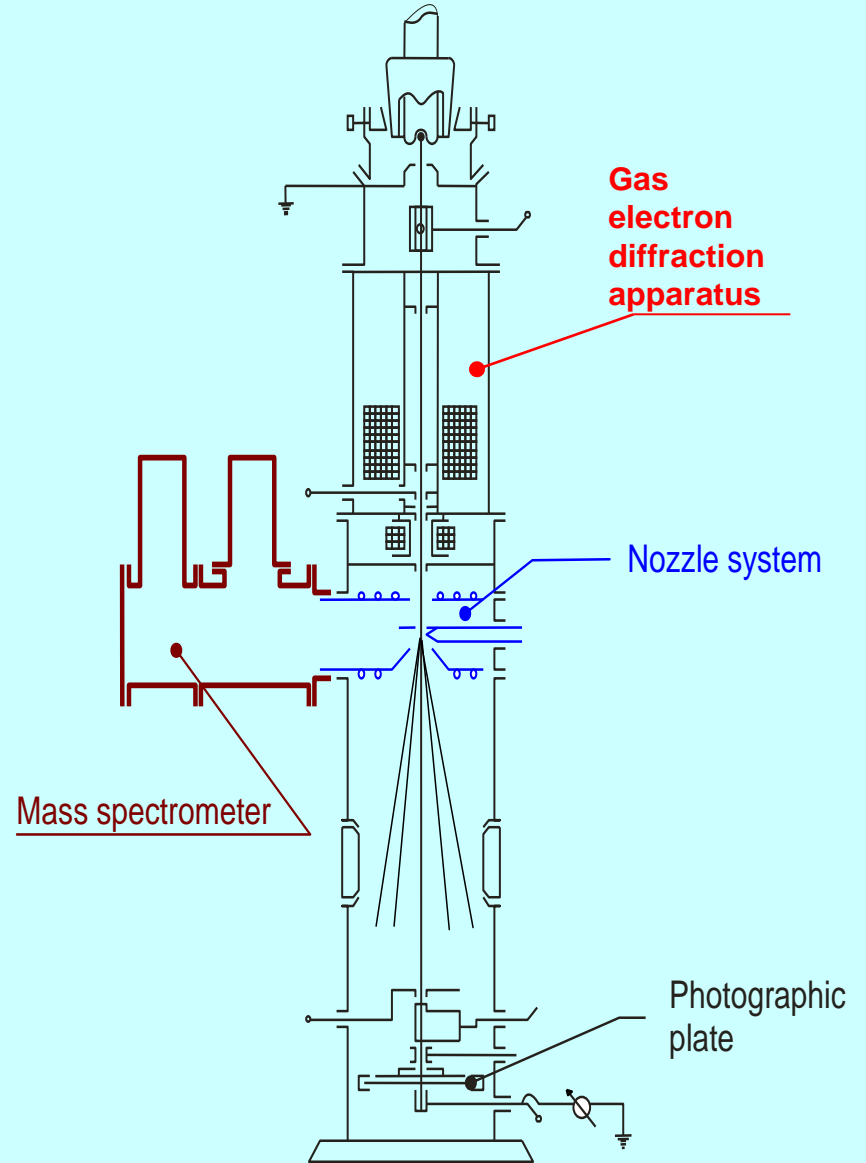
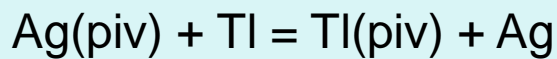
Experimental conditions

Effusion cell: graphite

Temperature: 425(5) K

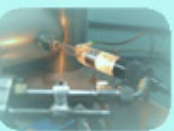
Wavelength of electrons: 0.0417 Å

The synthesis of thallium(I) pivalate was carried out *in situ*





Computational details



- Program : Gaussian 03
- Methods: DFT(B3LYP)
- Basis sets:
 - cc-pVTZ for **O**, **C** and **H**
 - ECP (60 electrons) + aug-cc-pVTZ-PP for **Tl**

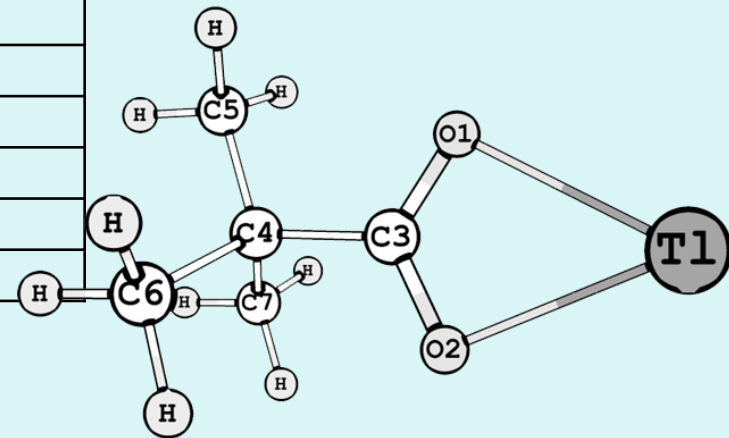
Structural parameters



Parameters ^a				
	r_e B3LYP/ cc-pVTZ	r_{hl}, \angle_{hl} GED	l GED	l QC
r(Tl-O1)	2.468	2.412(7)^B	0.143(4)	0.109
r(Tl-O2)	2.457	2.423(7)	0.146(4)	0.112
r(O1-C3)	1.266	1.263(4)	0.042(3)	0.043
r(O2-C3)	1.269	1.267(4)	0.042(3)	0.043
r(C3-C4)	1.539	1.533(4)	0.053(3)	0.055
r(C4-C5)	1.532	1.526(4)	0.053(3)	0.053
r(C4-C6)	1.540	1.534(4)	0.053(3)	0.055
r(C-H) ^b	1.091	1.096(5)	0.076(3)	0.077
r(O1...O2)	2.217	2.185(10)	0.086(4)	0.052
\angle O2-M-O1	53.5	53.7(2)		
\angle O2-C3-O1	122.0	119.5(6)		
\angle O1-C3-C4	119.9	120.6(6)		
\angle C3-C4-C5	111.1	110.7(4)		
\angle C3-C4-C6	108.0	107.6(4)		
τ (O1-C3-C4-C5)	0.0	8.0(80)		

^a Distances r , amplitudes l and shrinkage δ in Å, angle in deg.

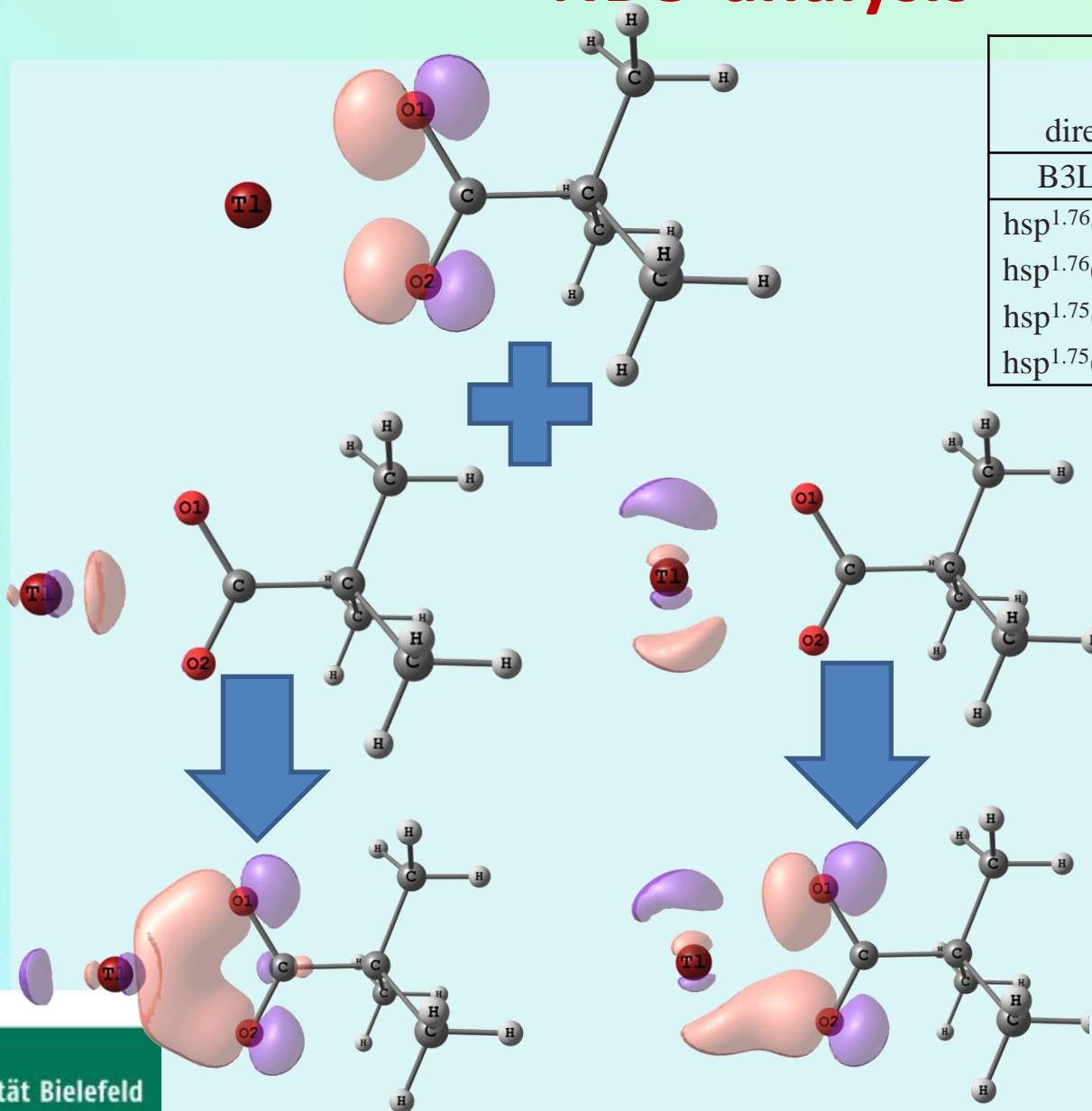
^b Uncertainties given in parentheses were taken as: $[(2.5\sigma_{LS})^2 + s^2_{scale}]^{1/2}$ ($\sigma_{scale} = 0.002r$, σ_{LS} is a standard deviation in least-squares refinement) for bond distances; $3\sigma_{LS}$ for bond angles.



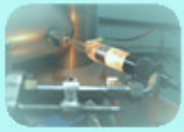
NBO-analysis

M←O	E ⁽²⁾	ΣE ⁽²⁾ (M←N)
direct donation		
B3LYP/ECP(Tl), cc-pVTZ(N,C,H)		
hsp ^{1.76} (O1)→ 6p _y (Tl)	10.89	17.37
hsp ^{1.76} (O1)→ 6p _x (Tl)	6.48	
hsp ^{1.75} (O2)→ 6p _y (Tl)	11.64	18.21
hsp ^{1.75} (O2)→ 6p _x (Tl)	6.57	

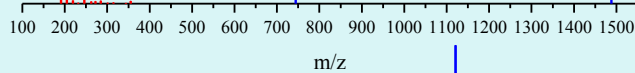
Second order perturbation energies E⁽²⁾ (kcal/mol) of orbital interactions in the thallium(I) pivalate



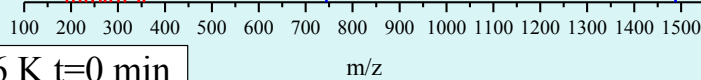
Magnesium Octa(m-trifluoromethylphenyl)porphyrazine



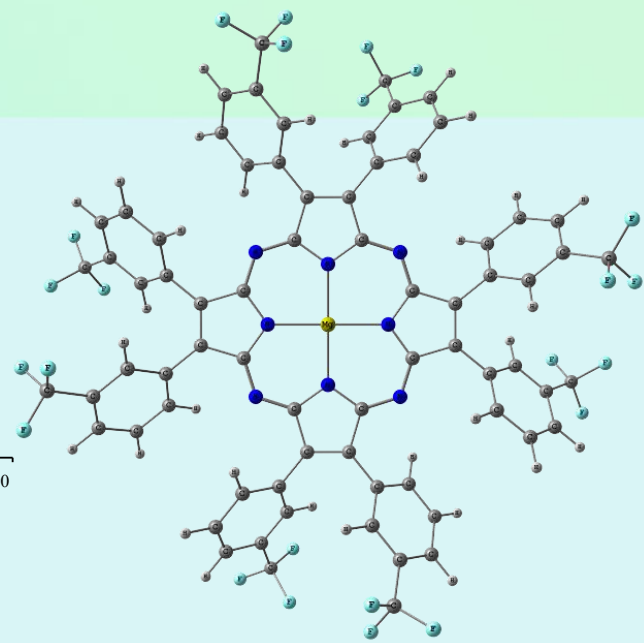
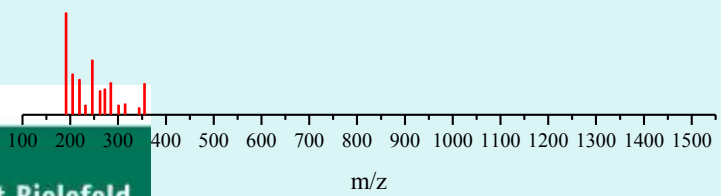
T= 668 K t=90 min



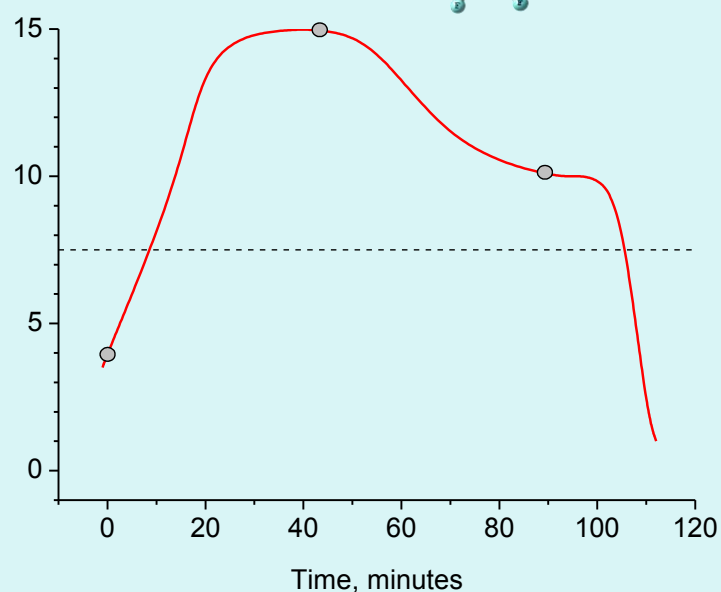
T= 568 K t=44 min



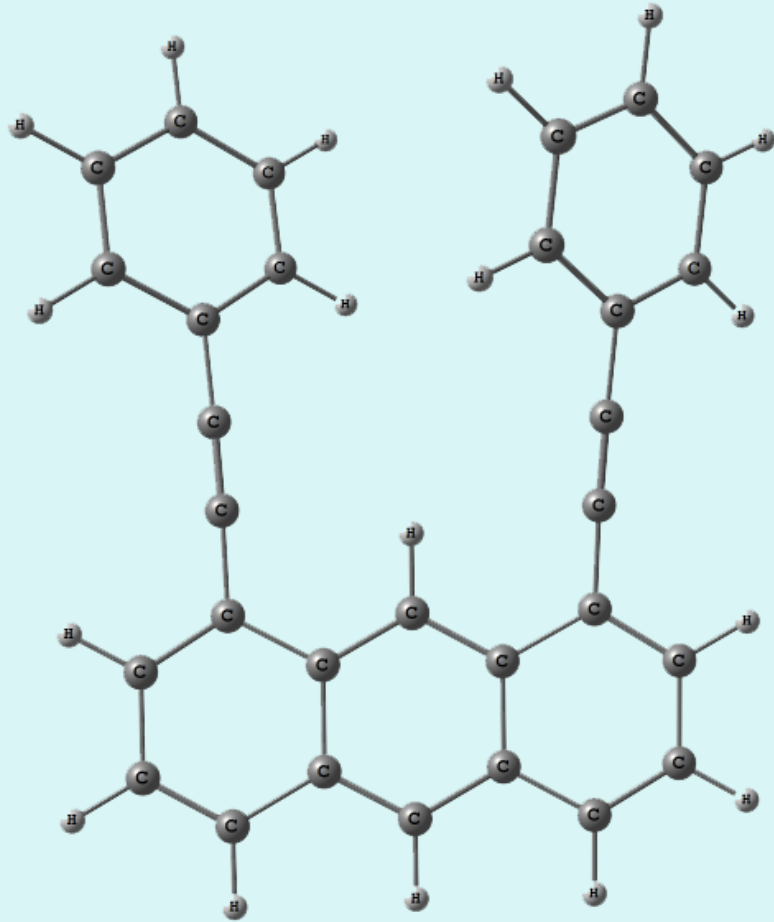
T=466 K t=0 min



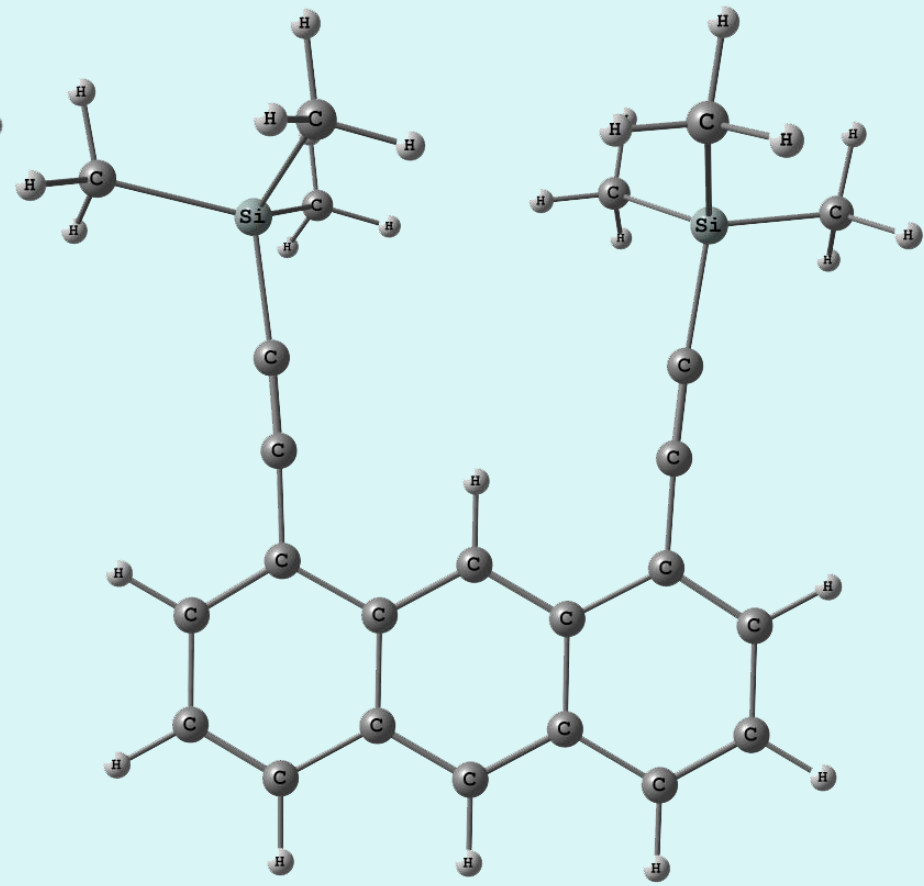
current of scattered electrons, relative units



Objects of DAAD project

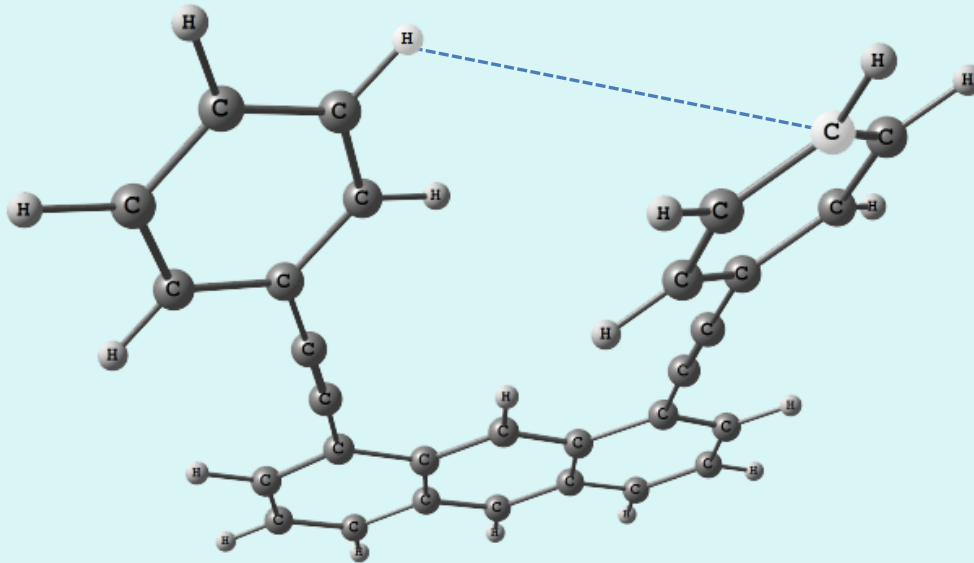


1,8-bis(phenylethynyl)anthracene



1,8-bis(trimethylsilylethynyl)anthracene

Problem of vibration corrections



$$r_e = 4.602 \text{ \AA}$$

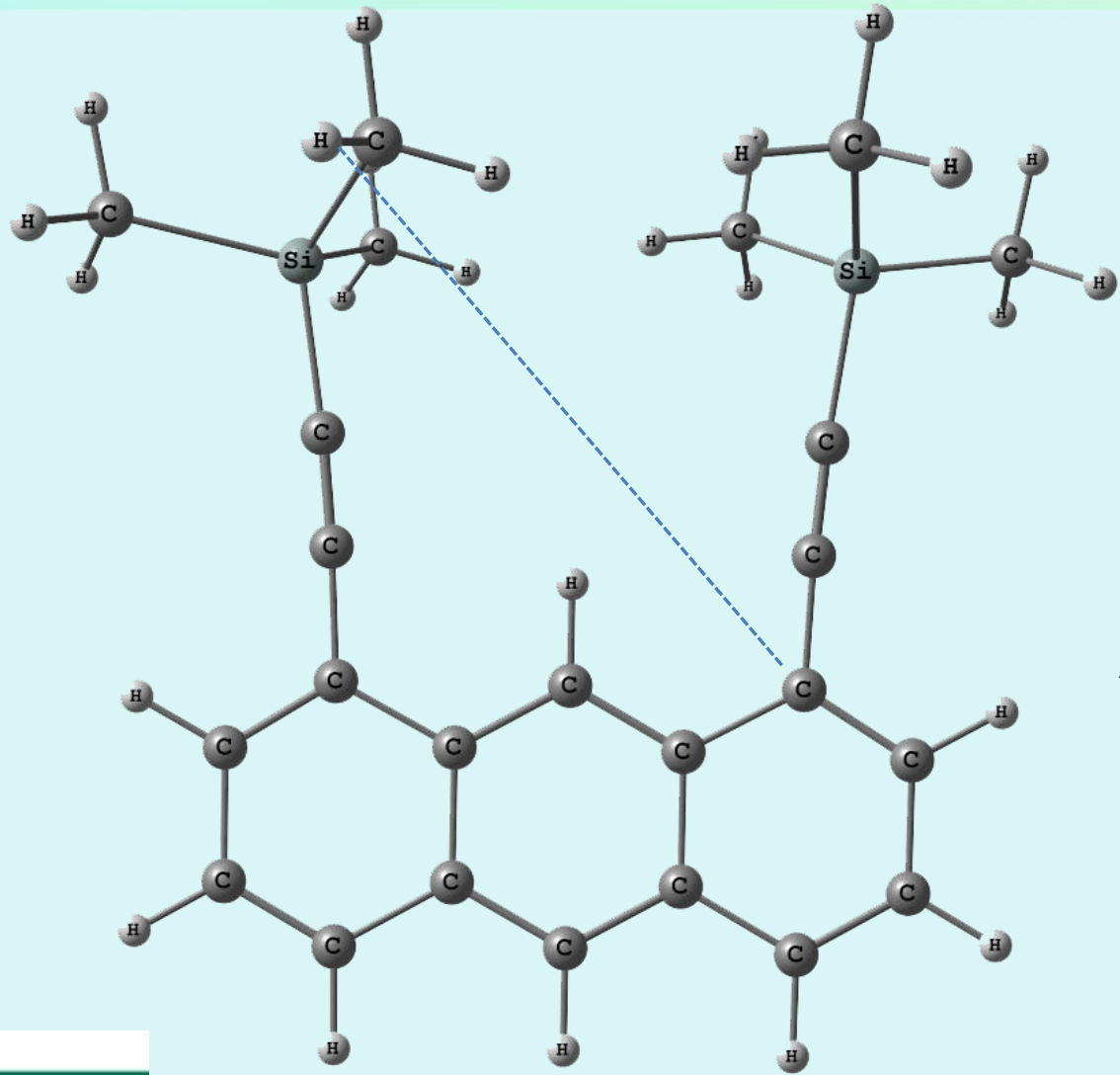
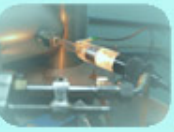
SHRINK

$$\Delta r = r_{h1} - r_a = 17.38 \text{ \AA}$$

MD simulation
cp2k
BLYP/DZVP

MDVibCor

$$\Delta r = r_e - r_a = -0.975 \text{ \AA}$$



$$r_e = 7.095 \text{ \AA}$$

SHRINK

$$\Delta r = r_{h1} - r_a = 9.616 \text{ \AA}$$

MDVibCor

$$\Delta r = r_e - r_a = -0.258 \text{ \AA}$$



Current state of research



Quantum chemistry calculations and structural analysis

Dr. Natalya Tverdova and stud. Arseniy Olyotov



MD simulation and calculation of vibration corrections

Dr. Yuriy Zhabanov





Thank you for attention !

