

Experimental and theoretical study of some aromatic molecules: results and challenges

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Seminar 29/05/2018

The objects:

Organic compounds

- ➤ acenaphthene
- > naphthalene, anthracene
- > rubrene
- > tris(pentafluorophenyl)(phenylethyl)silane
- Flexible bis-arenes: Ph-OC₂H₄O-Ph_f,
- Naph-OC₂H₄O-Ph_f, Anthr-OC₂H₄O-Ph_f

Organometallic compounds

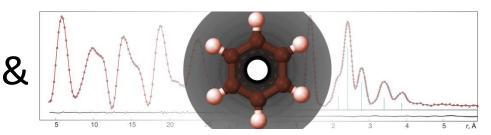
- perfluoro-ortho-phenylene mercury
- gold(I) dithiocarboxylate complex

The methods:



Gas-phase electron diffraction/
mass-spectrometry,
EMR-100/APDM-1, Ivanovo
(Prof. Dr. Georgiy V. Girichev,
Prof. Dr. Sergey A. Shlykov,
Dr. Vjacheslav M. Petrov)

EXPERIMENT



UNEX program (Dr. Yury V. Vishnevskiy)

REFINEMENT OF THE EXPERIMENTAL

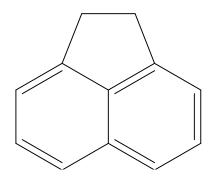
GED DATA

&

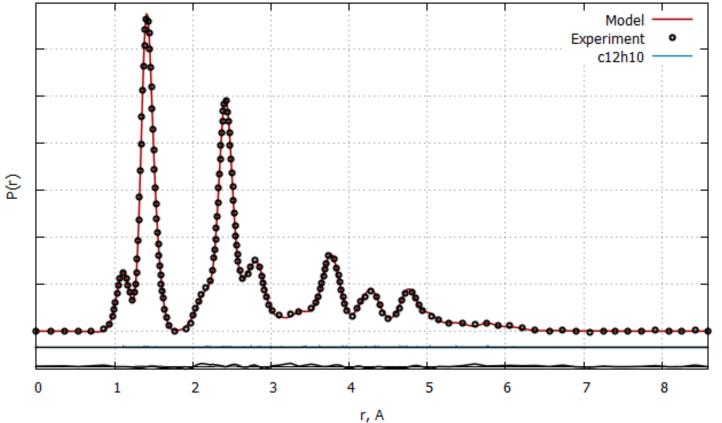
QUANTUM-CHEMICAL CALCULATIONS, including:

- Natural Bond Orbital analysis (NBO)
- Quantum Theory of Atoms In Molecules (QTAIM)

Acenaphthene

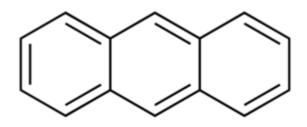


The first case of the **complete** refinement of GED experimental data from Ivanovo with use of UNEX program.

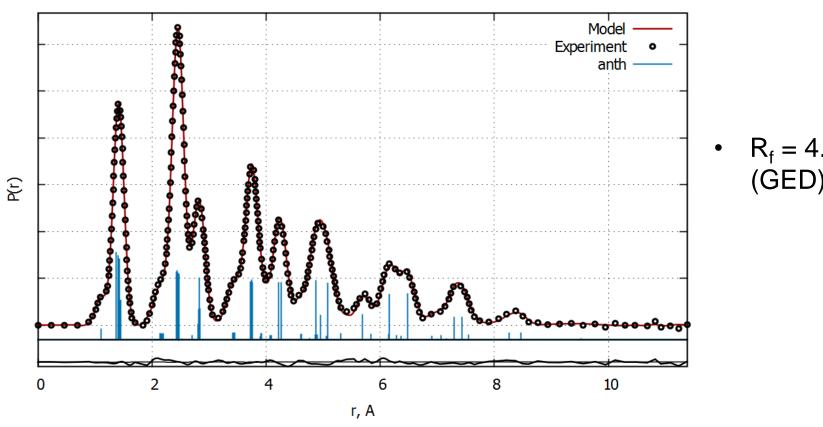


- $R_f = 3.92 \%$ (GED)
- $R_f = 3.86 \%$ (GED+MW)

Anthracene

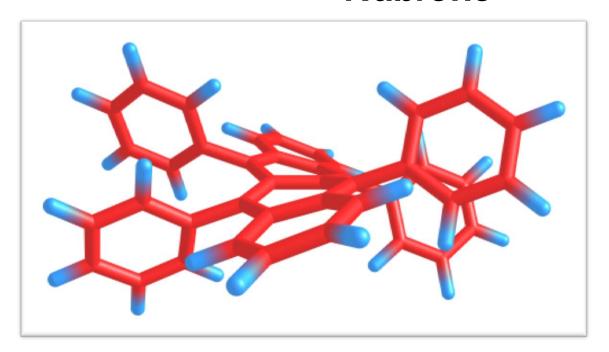


Very simple molecule, but challenging for the previous GED studies...



 $R_f = 4.36 \%$ (GED)

Rubrene

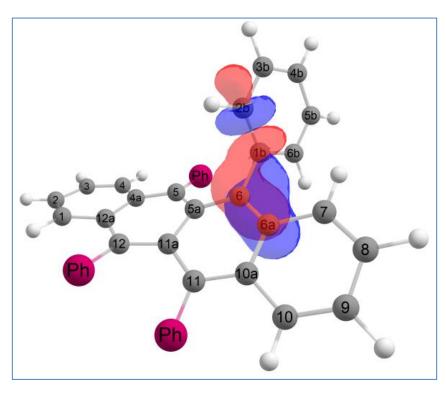


Equilibrium structure is *twist*-distorted

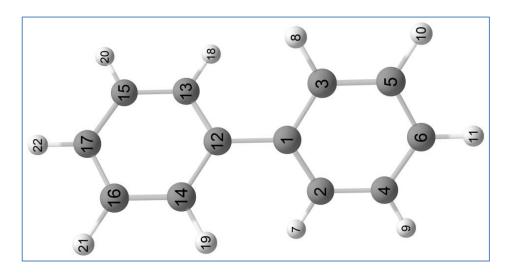
@ B3LYP/cc-pVTZ

Structure	$\mathbf{D}_{2\mathrm{h}}$	$\mathbf{C_{2h}}$	$\mathbf{D_2}$
ΔE, kJ mol ⁻¹	28.3	16.3	0.0
# of imaginary frequencies	2	1	0

NBO-analysis and comparison with Ph-Ph:

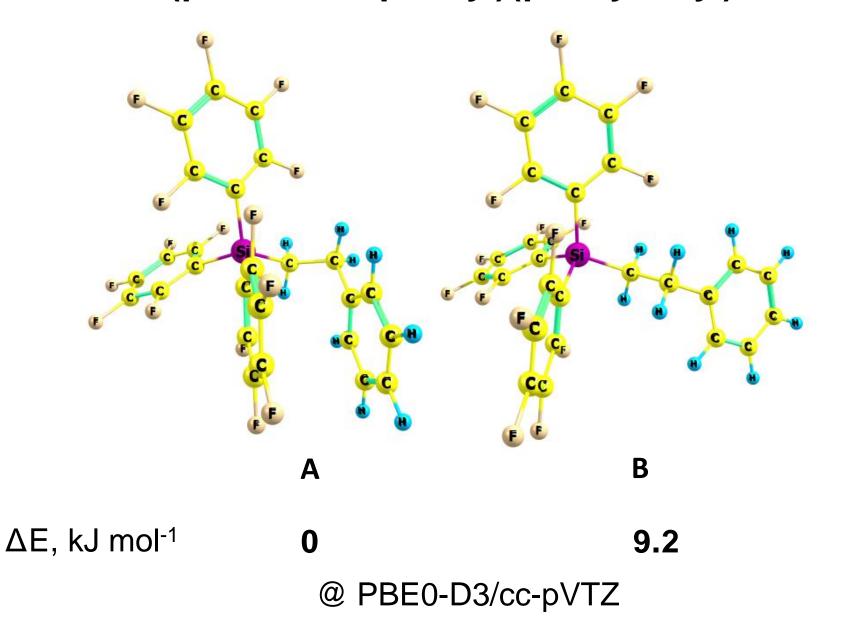


 $E^{(2)}_{sum} = 25.6 \text{ kcal mol}^{-1}$



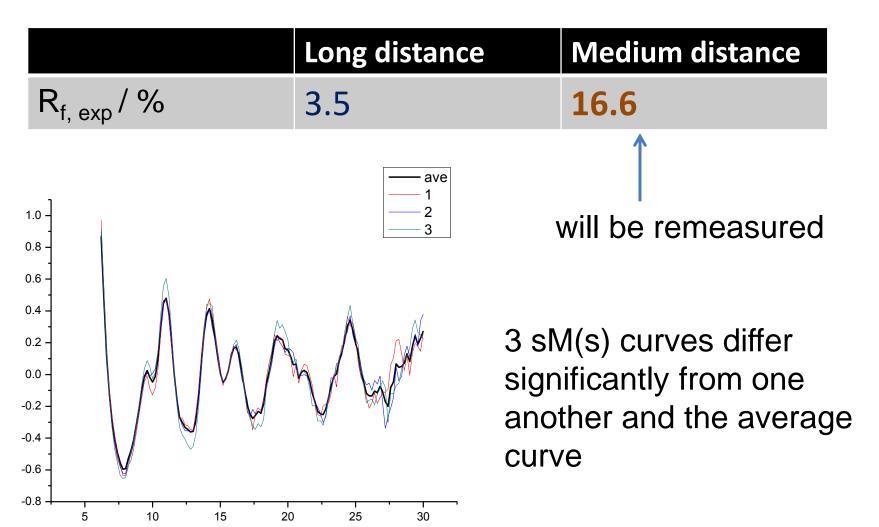
Ph-Ph			
τ/°	0°	40°	90°
E ⁽²⁾ _{sum} / kcal mol ⁻¹	42.4	34.6	22.1
ΔE / kcal mol ⁻¹	10.2	0.0	9.3

tris(pentafluorophenyl)(phenylethyl)silane

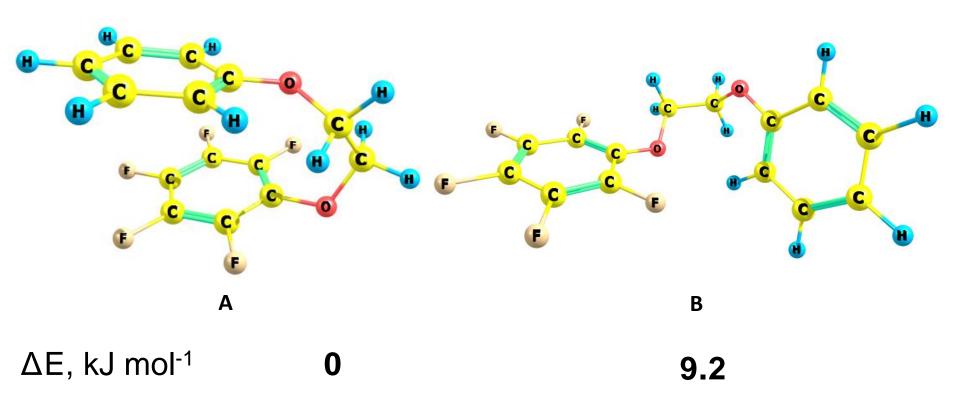


tris(pentafluorophenyl)(phenylethyl)silane

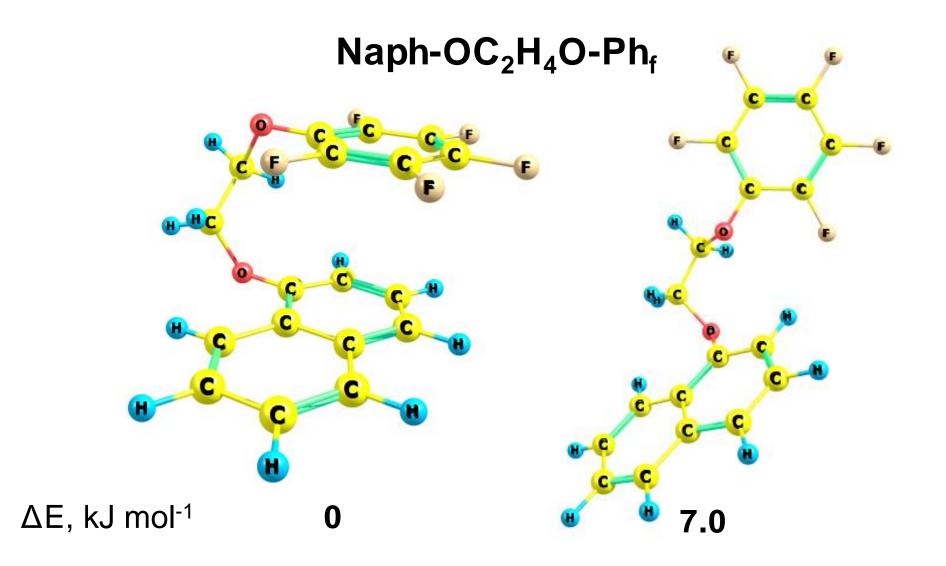
GED expirement, Bielefeld



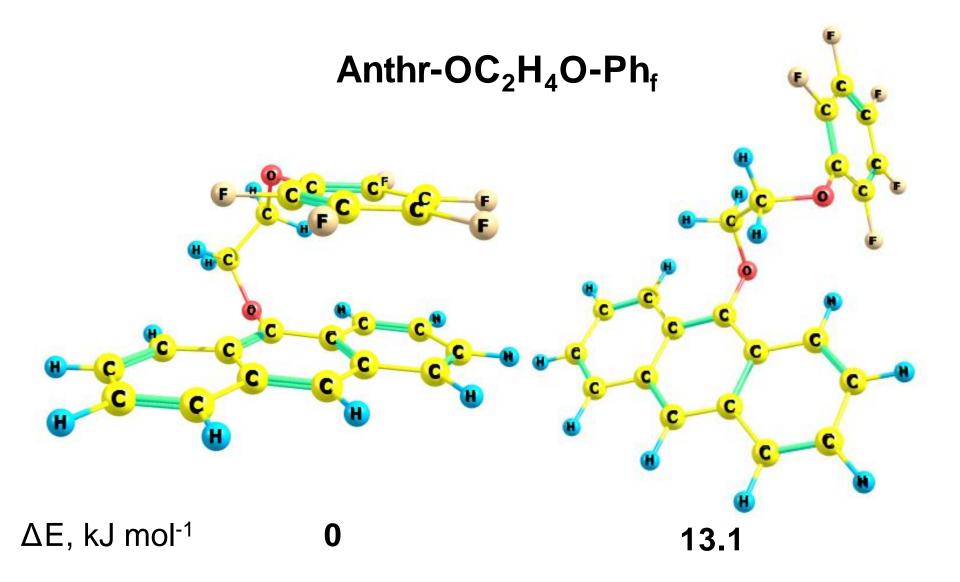
1-(pentafluorophenoxy)-2-(phenoxy)ethane (Ph-OC₂H₄O-Ph_f)



@ PBE0-D3/cc-pVTZ

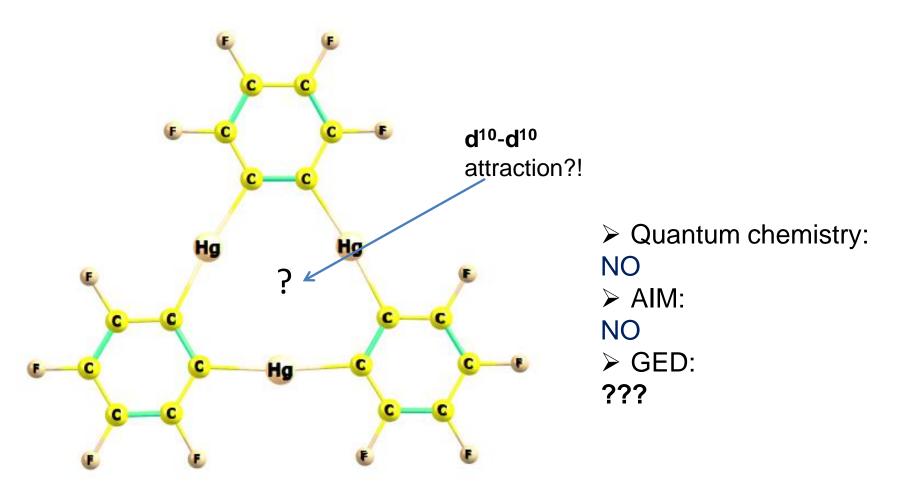


@ PBE0-D3/cc-pVTZ



@ PBE0-D3/cc-pVTZ

(perfluoro-orho-phenylene) mercury



Problems of the GED refinement:

	Long distance	Short distance
R _{f, exp} / %(photometric treatment)	25.7	54.6
R _{f, exp} / %(scanner treatment)	13.8	15.1

these very large values make a successful refinement impossible from the beginning...

What is the reason of these disagreements within the sets of the experimental data?

Vapor composition during the GED/MS experiment

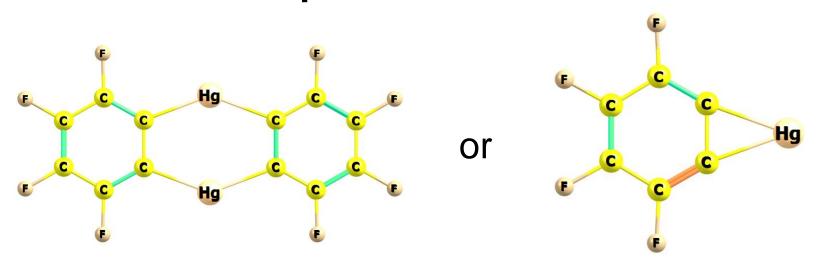


Mass-spectrum is highly reproducible

Any products of thermal decomposition in the vapor?

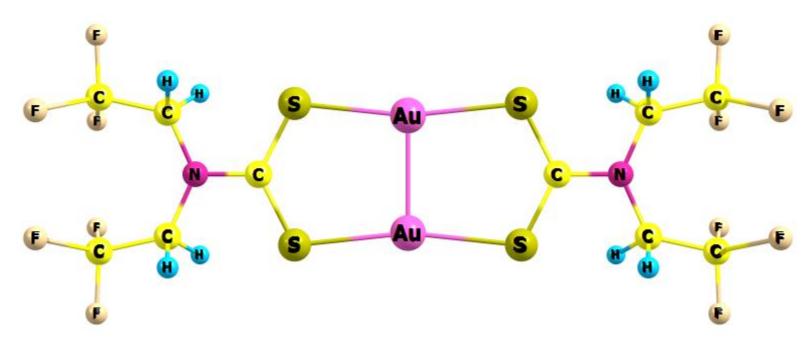


incompleteness of the model

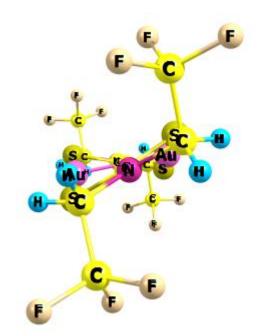


including of these species to the model did not result in improvement of the R_f

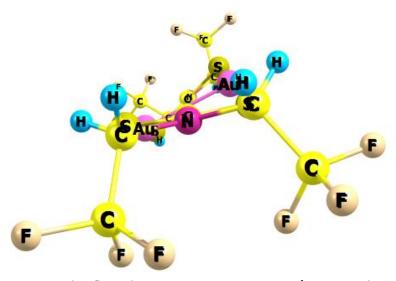
Gold(I) dithiocarboxylate complex



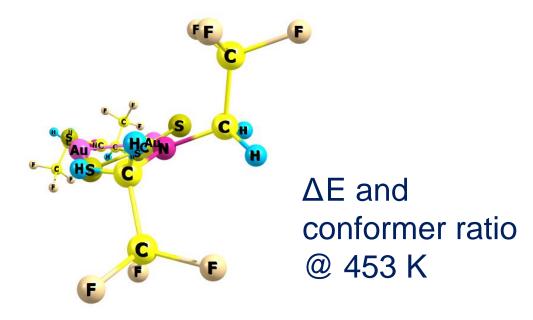
Another promising candidate for **d**¹⁰-**d**¹⁰ attraction in the gas phase



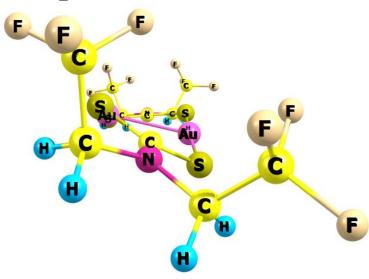
1) **D**₂ (0 kcal mol⁻¹; 36 %)



3) **C**₂ (7.87 kcal mol⁻¹; 0 %)



2) **C**₂ (0.54 kcal mol⁻¹; 64 %)



4) **C**₂ (8.00 kcal mol⁻¹; 0 %)

Molecule(s)	What has been done	Current state
Acenaphthene	GED,GED+MW, QC (+NBO), MS	manuscript in preparation
Naphthalene, anthracene	GED (Ivanovo), QC (+NBO)	GED+MW refinement in progress; GED (Bielefeld) is to follow
Rubrene	QC (+NBO)	GED/MS (Ivanovo) and GED (Bielefeld?!) is to follow
tris(pentafluorophenyl) (phenylethyl)silane	QC, preliminary treatment of the GED data	GED (Bielefeld, MD) is to follow
$Ph-OC_2H_4O-Ph_f;$ $Naph-OC_2H_4O-Ph_f;$ $Anthr-OC_2H_4-Ph_f$	QC	GED (Ivanovo) is to follow

Molecule(s)	What has been done	Current state
perfluoro- <i>ortho</i> -phenylene mercury	GED/MS (failed), QC (+AIM)	MS study at different U _{ion} is to follow
Gold (I) dithiocarboxylate	GED (Bielefeld), QC (preliminary results)	GED refinement of the 2-conformer mixture is in progress

Thank you for your attention!