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## **Low-Pressure Gas Electron Diffraction**

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16<sup>th</sup> ESGED, Frauenchiemsee, June 21 – 26, 2015

GED for low-volatile and thermally unstable compounds.

 $\begin{array}{ll} \text{Normal GED: } P_{\text{sample}} \sim 1 - 10 \text{ mbar} \\ \text{LP GED:} & P_{\text{sample}} < 10^{\text{-2}} \text{ mbar} \end{array}$ 

- High-temperature GED.
- Combined MS+GED experiments.
- Molecular beams.
- Highly reproducible experimental conditions.
- Experimental background.

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#### Molecular beams

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C. B. Lucas, Atomic and Molecular Beams, CRC Press, 2014.

Proposed by A. A. Ivanov (MSU).

Target — "Molecular film".



#### CAD Model



## Inlet system



# Real prototype







MS

#### MS+GED @ Bielefeld



Hiden EPIC ion counting detector (up to 2500 amu)

#### **Test measurements**

#### First measurements: Benzoic acid

	Vishnevskiy et al., 2015	Aarset et al., 2006
Temperature, K	287	406
Sample pressure, mbar	2.7×10 <sup>-4</sup>	17
Background pressure, mbar	1×10 <sup>-7</sup>	?
Electron beam current, µA	10.9	?
Exposure time, s	60	?



K. Aarset, E. M. Page, D. A. Rice, J. Phys. Chem. A, 110, 2006, 9014.
M. Colomina, P. Jimenez, C. Turrion, J. Chem. Thermodyn., 14, 1982, 779.
C. G. de Kruif, J. G. Blok, J. Chem. Thermodyn., 14, 1982, 201.

#### Benzoic acid

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Parameter, [Å, °]	Vishnevskiy et al., 2015	Aarset et al., 2006
$r_{\rm a}({\rm C_{ar}-C_{ar}})_{\rm av}$	1.393(1)	1.397(6)
$r_{\rm a}({\rm C_{ar}}-{\rm C})$	1.475(7)	1.475(21)
$r_{a}(C=O)$	1.207(4)	1.220(18)
$r_{\rm a}$ (C-O)	1.347(8)	1.359(24)
$l(C_{ar}-C_{ar})$	0.053(2)	0.054(18)
R <sub>f</sub> , %	2.6	?

Errors:  $3\sigma$ 

K. Aarset, E. M. Page, D. A. Rice, J. Phys. Chem. A, 110, 2006, 9014.

# Iodoform, CHI<sub>3</sub>

	Vishnevskiy et al., 2015	Takeuchi et al., 2003
Temperature, K	288	395
Electron beam current, $\mu A$	0.9	1.5
Exposure time, s	45 - 90	110 - 140



H. Takeuchi, T. Ozaki, T. Takeshima, T. Egawa, S. Konaka, J. Mol. Struct., 657, 2003, 381.

## Iodoform, CHI<sub>3</sub>

1...1

4

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Parameter,	Takeuchi et al., 2003	Vishnevskiy et al., 2015		Calculated
[A, <sup>-</sup> ]	$\Gamma_{\rm g}$	Γ <sub>g</sub>	Γ <sub>e</sub>	Γ <sub>e</sub>
<i>r</i> (C-H)	1.111 assumed	1.128(107)	1.107(107)	1.081
<i>r</i> (C–I)	2.145(8)	2.130(6)	2.119(6)	2.125
<i>r</i> (II)	3.549(2)	3.536(1)	3.522(1)	3.531
l(C-I)	0.067(7)	0.0	67(1)	0.055
<i>l</i> (II)	0.108(2)	0.101(2)		0.084
R <sub>f</sub> , %	?	Ę	5.1	

QC geometry: CCSD(T)/CBS-PP, Amplitudes: MP2/SDB-cc-pVTZ

Errors:  $3\sigma$ 

Δ

6

5

H. Takeuchi, T. Ozaki, T. Takeshima, T. Egawa, S. Konaka, J. Mol. Struct., 657, 2003, 381.

## Carbon tetraiodide, $CI_4$

	Vishnevskiy et al., 2015	Hargittai et al., 2001
Temperature, K	290	396
Electron beam current, µA	4.8	?
Exposure time, s	20 - 60	?

#### Combined GED+MS @ 290 K



Hargittai: ~ 20%  $I_2$  @ 396 K



**Fig. 2.** Experimental (E) and calculated (T) radial distributions for Model 1 and their differences ( $\Delta$ ) for the three models of Table VII. The contribution of the distances of CI<sub>4</sub> and I<sub>2</sub>, and the position of the longest I···I distance of C<sub>2</sub>I<sub>4</sub> from Model 1 are indicated.  $\Delta$ 4 corresponds to a model with CI<sub>4</sub> and I<sub>2</sub> only.

M. Hargittai, G. Schultz, P. Schwerdtfeger, M. Seth, Struct. Chem., 12, 2001, 377.

#### Carbon tetraiodide, $CI_4$





Parameter,	Hargittai et al., 2001	Vishnevskiy et al., 2015		Calculated
[Å, °]	$r_{ m g}$	$r_{ m g}$	$r_{ m e}$	$r_{ m e}$
r(C-I)	2.157(6)	2.142(1)	2.132(1)	2.144
<i>r</i> (II)	3.530(7)	3.496(2)	3.482(2)	3.501
<i>l</i> (C–I)	0.063 assumed	0.046(22)		0.058
<i>l</i> (II)	0.106(23)	0.114(3)		0.083
R <sub>f</sub> , %	5.4	7.1		

QC geometry: CCSD(T)/CBS-PP, Amplitudes: MP2/SDB-cc-pVTZ

Errors:  $3\sigma$ ,  $2\sigma$ +se

M. Hargittai, G. Schultz, P. Schwerdtfeger, M. Seth, Struct. Chem., 12, 2001, 377.

- Ring cell works!
- Successfully tested MS+GED.
- It is possible to avoid substance decomposition.
- Refined parameters are
  - a) precise and accurate,
  - b) consistent with published data.

- Implement heating of the effusion cell.
- Tune mass-detector and optimize its position.
- Measure new compounds.

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- Christian G. Reuter
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#### DFG

#### Thank you for your attention!